Science English

Analyze the Results "Tatara iron making"

Self – Introduction & Free Conversation with TA

Today's word

Today's Words

- conduct (熱、電気)伝える
- electric current 電流
- metallic luster 金属光沢
- copper 銅
- oxidation-reduction reaction 酸化還元反応
- confirm 確かめる
- combine 結びつける

Today's Words

- matter 物質
- product 生成物
- oxide 酸化物
- observation 観察
- experiment 実験
- describe ~の特徴を述べる
- apply 適用・応用・利用する

Conduct (熱、電気) 伝える

conduct electricity / conduct heat

Current 電流

An *electric current* runs through this wire.

metallic luster 金属光沢



COPPer 鋼



Copper is a metal.

oxidation-reduction reaction with reaction 酸化還元反応

oxidation $4HC1 + O_2 \longrightarrow 2Cl_2 + 2 H_2O$ reduction

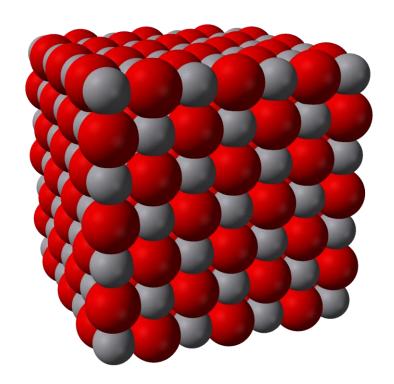
Confirm 確かめる



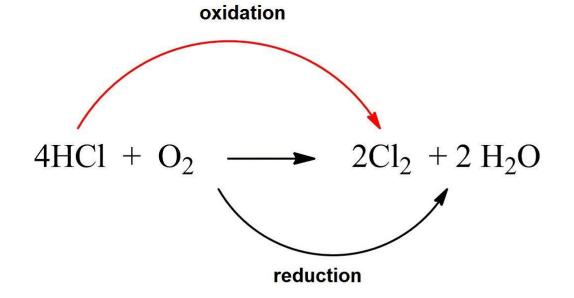
She is *confirming* the safety of the window.

Combine 結びつける

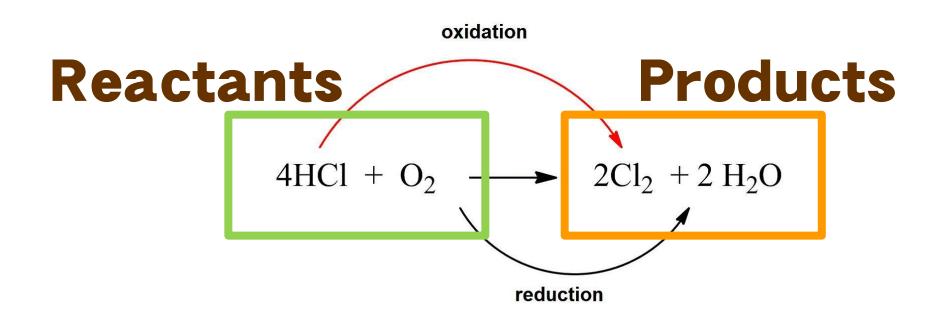
matter 物質



oxidation-reduction reaction with reaction 酸化還元反応



product 生成物



OXICE 酸化物



observation 観察

He *observes* flowers every day.

experiment 実験

Mendel used pea plants in his experiments.

describe ~の特徴を述べる

Describe the characteristic of girl A.

apply 適用・応用・利用する

The principles of electromagnetic induction are *applied* in many devices.

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Vocabulary games with TA

Science English

Analyze the Results "Tatara iron making"

今日のポイント

Scientific Approach

Make an Observation Think why?, how?

Do Background Research

1たたら製鉄とは? (化学分野の勉強)

Construct a Hypothesis

仮説の設定

Design a Science Approach 科学的アプロ 一チの方法

Test with an Experiment

実験

Analyze the Results

結果の分析

2この過程で < 何が大切か? (研究手法の勉強)

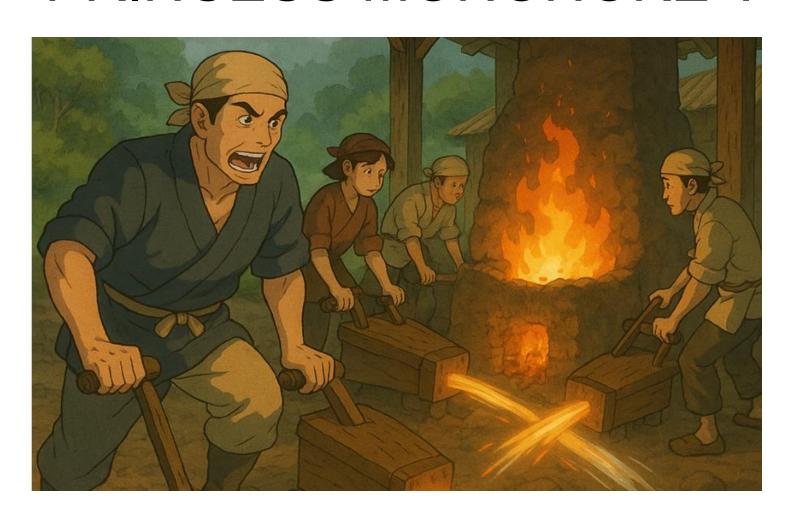
Discussion

考察

Draw Conclusions

結論

Have you ever watched "PRINCESS MONONOKE"?



Tatara iron making



Ores

Metal oxidesMetal sulfides

Hematite

(Fe₂O₃)赤鉄鉱



Cassiterite

(SnO) 錫石









(PbS)

Galena

Iron sand (Fe₂O₃) 砂鉄

Peacock ore



Microwave oven



CuO



Science Approach

Aim 目的

To develop a method to take out metal from ore at a laboratory level. 実験室での方法を開発する

● What is known わかっていること

Tatara iron making



We can heat up a charcoal



Hypothesis 仮説

Copper will be taken out from copper oxide by heating the oxide with charcoal powder using a microwave oven.

電子レンジの加熱でクジャク石から銅を取り出せる

malachite

ceramic dish









Mechanism: Metal Oxide → Metal

metal oxide









CO

charcoal

carbon monoxide

charcoal



$Fe_2O_3+2CO \rightarrow 2Fe+3CO_2$

lose oxygen

Fe₂O₃ reduced 還元される

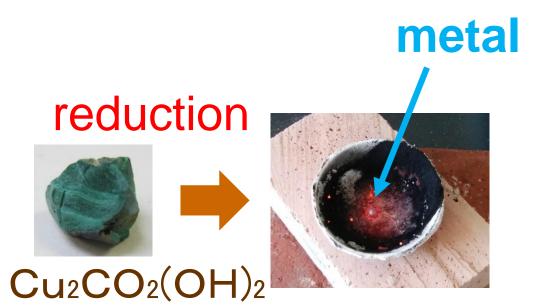


Decompose 分解する

 $Cu_2CO_3(OH)_2 \longrightarrow 2CuO + CO_2 + H_2O$

 $2\underline{CuO} + \underline{C} \longrightarrow 2\underline{Cu} + \underline{CO_2}$

CuO reduced 還元される





Critical Thinking 批判的思考

Is it true?
Is this really a metal?



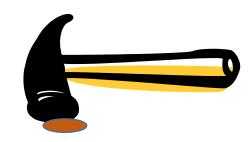
Metal?
Copper?

What is the property of all metals? 特性

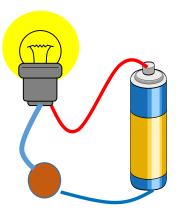
What is the property of all metals? 特性







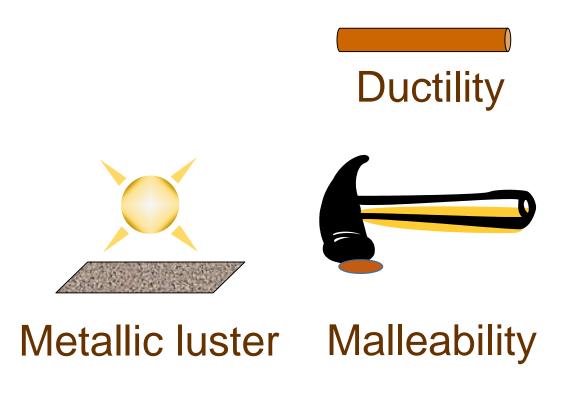
Malleability

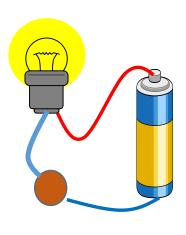


Electrical conductivity

Discuss with TA

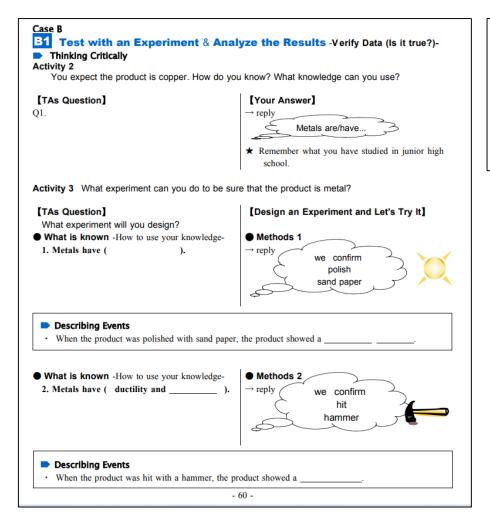
How do you confirm whether this matter is a metal or not?

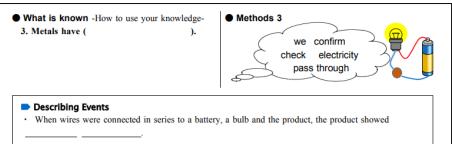




Electrical conductivity

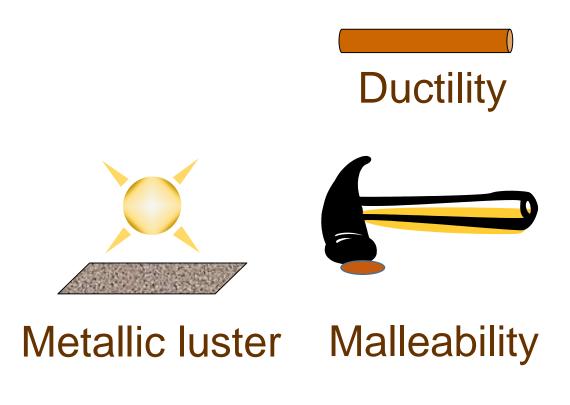
Discuss with TA

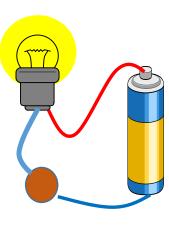




Experiment with TA

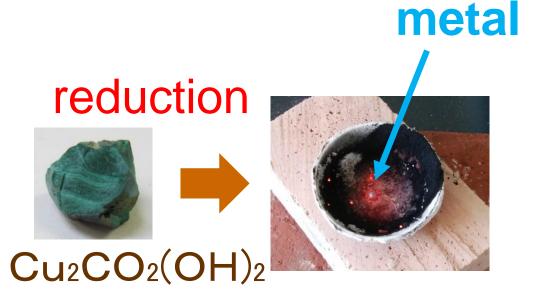
How do you confirm whether this matter is a metal or not?





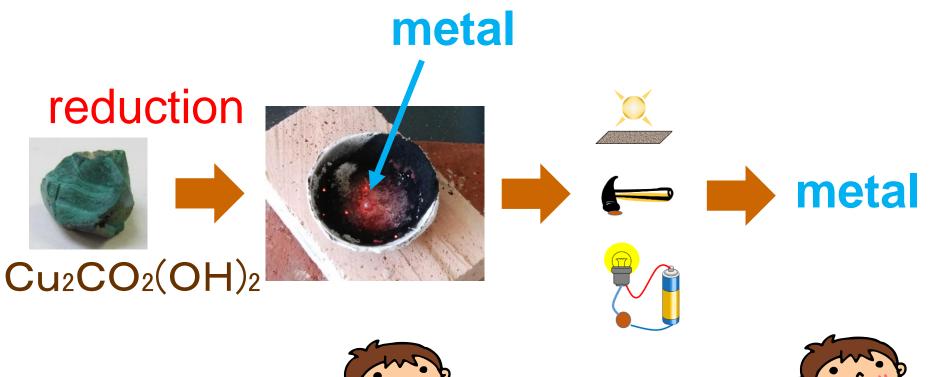
Electrical conductivity

Case A: Finish your experiment when you got the product.





Case B: Finish your experiment when you checked the product for the three properties of metal.





success



success

Activity 4

In a scientific research, you conduct an experiment to identify the hypothesis that "Copper will be taken out from minerals by applying oxidation-reduction reactions". Discuss with your group what is the difference between the cases below. Then write down the best single English word which represents the difference.

Case A: Finish your experiment when you got the product.

生成物を得て実験を終える

Case B: Finish your experiment when you checked the product for the three properties of metal.

金属の3特性について生成物をチェックして実験を終える

Case C: Finish your experiment when you identified the product is copper.

生成物が銅であることを確認して実験を終える

Case C has more

reliability 信輔性

than Case A in order to conclude that the product is copper.

Presentation Exercise

Smelting











Ores

Metal oxidesMetal sulfides

Hematite

(Fe₂O₃)赤鉄鉱



Cassiterite

(SnO) 錫石







Iron sand (Fe₂O₃) 砂鉄

Mechanism: Metal Oxide → Metal

metal oxide









CO

charcoal

carbon monoxide

charcoal



Fe₂O₃+2CO → 2Fe+3CO₂

lose oxygen

Fe₂O₃ reduced 還元される



Decompose **分解する**

 $Cu_2CO_3(OH)_2 \longrightarrow 2CuO + CO_2 + H_2O$

 $2\underline{CuO} + \underline{C} \longrightarrow 2\underline{Cu} + \underline{CO_2}$

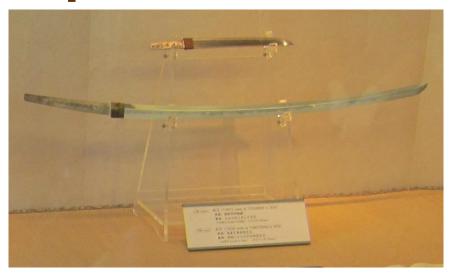
CuO reduced 還元される

"Tatara iron making"

Edo period



Japanese swords



→Starting point of Japanese manufacturing technology →

Thank you very much!